

## THE FORMATION OF 2,3-DIHYDRO-2,3-DIHYDROXY AFLATOXIN B<sub>1</sub> BY THE METABOLISM OF AFLATOXIN B<sub>1</sub> IN VITRO BY RAT LIVER MICROSOMES

G. E. NEAL<sup>+</sup> and P. J. COLLEY\*

*Toxicology Unit, MRC Laboratories, Carshalton, Surrey, SM5 4EF, England*

Received 26 March 1979

### 1. Introduction

The theory has been advanced that aflatoxin B<sub>1</sub> (AFB<sub>1</sub>) is activated by microsomal systems to form aflatoxin B<sub>1</sub> 2,3-oxide [1,2]. This reactive metabolite has not been isolated and might be expected to be rapidly hydrolysed, either enzymically or nonenzymically, to form 2,3-dihydro-2,3-dihydroxy aflatoxin B<sub>1</sub> (AFB<sub>1</sub>-dhd). However, it has been reported that only very low levels of AFB<sub>1</sub>-dhd are formed by metabolism of AFB<sub>1</sub> in rat liver microsomal incubations at pH 6.5 (2.9% conversion of AFB<sub>1</sub> using microsomes from phenobarbitone pre-treated rats and 0.6% using control microsomes) and that even smaller amounts of AFB<sub>1</sub>-dhd are formed at physiological pH [3]. The latter finding was ascribed to the destruction of AFB<sub>1</sub>-dhd at neutral pH. We here present evidence that AFB<sub>1</sub>-dhd is a major metabolite of AFB<sub>1</sub> produced by rat liver microsomes at physiological pH but it is removed by binding to microsomal protein. This binding does not occur at pH 6.5. Amines including the buffer tris-hydroxymethyl amino methane (Tris) and both oxidised and reduced glutathione can compete with microsomal protein for binding AFB<sub>1</sub>-dhd. Changes in the ultraviolet absorption spectrum of AFB<sub>1</sub>-dhd indicating the formation of a dialdehydic phenolate ion correlate with the binding to amines and microsomal proteins at neutral pH and indicate that binding proceeds via the Schiff's base reaction previously suggested for the binding of the hemiacetal aflatoxin B<sub>2a</sub>.

\* Present address: Laboratory Data Control, Milton Roy House, High Street, Stone, Staffs, England

<sup>+</sup> To whom correspondence should be addressed

### 2. Materials and methods

#### 2.1. Chemicals

AFB<sub>1</sub> was obtained from Makor Chemical Co., Jerusalem. G-<sup>[3H]</sup>AFB<sub>1</sub> spec. act. 45 Ci/mmol was obtained from Moravsek Biochemicals Ltd, City of Industry, CA. AFB<sub>1</sub>-dhd was prepared by reaction of AFB<sub>1</sub> with osmium tetroxide [5], <sup>[3H]</sup>AFB<sub>1</sub>-dhd was prepared by the same method using <sup>[3H]</sup>AFB<sub>1</sub> as the starting material. The preparation of other aflatoxins was as in [6].

Tris, glycine, and oxidised and reduced glutathione were AR grade obtained from Sigma (London) Chemical Co., Ltd, Poole, Dorset. Other chemicals and solvents used were supplied by the sources in [6].

#### 2.2. AFB<sub>1</sub> metabolism

Microsomes, prepared from the livers of adult male F344/TIF Lac (Fischer) rats given 0.1% phenobarbitone in the drinking water for 3–8 days, were incubated with AFB<sub>1</sub> and cofactors at 37°C in O<sub>2</sub> with shaking [6]. The buffer used was either phosphate or Tris–HCl (both were 80 mM, pH 7.4). All procedures were carried out in subdued light. Metabolism was stopped after 30 min by the addition of 2 ml ice-cold methanol. After 30 min in ice the incubate was centrifuged at 2000 rev./min and 4°C. The pellet was washed with 2 ml ice-cold methanol and centrifuged. The combined supernatants were blown to dryness at 50°C under a stream of O<sub>2</sub>-free N<sub>2</sub>. The residue was dissolved in 2 ml of methanol/water (1:1, v/v), centrifuged for 30 min at 2000 rev./min and –10°C to clarify and used for HPLC analysis.

#### 2.3. High-performance liquid chromatography (HPLC)

A Du Pont model 830 liquid chromatograph, 833

flow controller, 834 automatic sampler (fitted with a 100  $\mu$ l injection loop) and 837 variable wavelength spectrophotometer (operated at 365 nm) were used. The fluorescence of the column eluates was monitored using a Kontron Spectrofluorometer SFM 23 fitted with an 8  $\mu$ l flow cell. The column used was a Partisil-10-ODS (250  $\times$  4.6 mm) obtained from Whatman LabSales Ltd (Maidstone) Kent, and results were displayed on a Hewlett-Packard model 3380S recording integrator. Elution of AFB<sub>1</sub> and its metabolites was achieved using 15% dimethylformamide in 0.01% phosphoric acid at 1.5 ml min<sup>-1</sup> and a column temperature of 69°C [7].

#### 2.4. Thin-layer chromatography (TLC)

Silica gel G thin-layer plates (0.25 mm thick) were obtained from Anachem Ltd, Luton, Bedfordshire and were used without activation. Plates were developed in either chloroform:acetone (9:1, v/v) or chloroform:methanol (9:1, v/v) and spots visualized by their fluorescence under long wavelength ultraviolet light.

### 3. Results and discussion

When aflatoxin B<sub>1</sub> was metabolised by rat liver microsomes in either phosphate or Tris buffer and the metabolites subjected to HPLC analysis and the ultraviolet absorption of the eluate monitored, the chromatograms shown in fig.1a,b were obtained. The only significant difference between these chromatograms is that a major product having a retention time of ~6 min was present in the Tris incubations but was absent from the phosphate incubations. This compound had previously been identified as a degradation product (Tris-diol) obtained when authentic AFB<sub>1</sub>-dhd was treated with neutral Tris buffer [6]. The absence of AFB<sub>1</sub>-dhd from the phosphate incubations was confirmed by the parallel fluorescence monitoring of the HPLC eluate (excitation (Ex) 372 nm; emission (Em) 446 nm) (fig.1c,d). AFB<sub>1</sub>-dhd and its Tris degradation product are more highly fluorescent than aflatoxin B<sub>1</sub> or the other aflatoxin metabolites detected in these studies (fig.1a,c). Despite the absence of the Tris-diol peak when AFB<sub>1</sub> was incubated in phosphate buffer, the overall amounts of AFB<sub>1</sub> remaining unmetabolised and the rates of formation

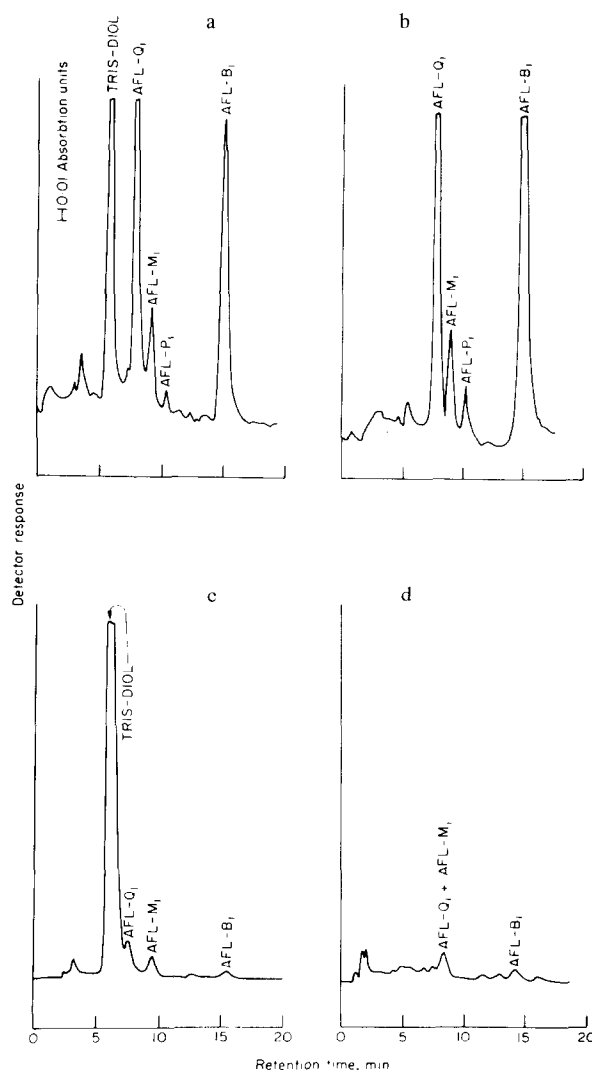


Fig.1. HPLC separations of microsomal metabolites of AFB<sub>1</sub>. Incubations, preparations of samples and HPLC carried out as detailed in section 2. (a) Incubations carried out in presence of Tris buffer. Ultraviolet absorption of HPLC effluent monitored at 365 nm. (b) Incubations carried out in the presence of phosphate buffer. Ultraviolet absorption monitored at 365 nm. (c) Incubation carried out in presence of Tris buffer. Fluorescence of HPLC effluent monitored at Ex 372 nm Em 446 nm. (d) Incubation carried out in presence of phosphate buffer. Fluorescence monitored at Ex 372 nm Em 446 nm. Fluorescence measurements at Ex 372 nm Em 446 nm optimal for AFB<sub>1</sub>-dhd and Tris-diol. Optima for aflatoxins B<sub>1</sub>, M<sub>1</sub> and Q<sub>1</sub> Ex 365 nm Em 425, but relative fluorescence of these aflatoxin metabolites at optimal settings essentially as in (c).

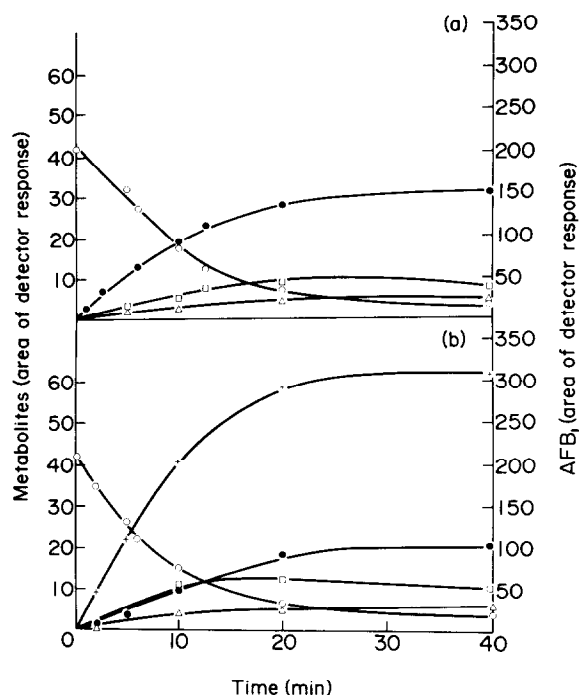


Fig.2. Rate of microsomal metabolism of AFB<sub>1</sub> and production of metabolites in phosphate and Tris buffer. Incubations, preparation of samples and HPLC carried out as detailed in section 2. (a) Phosphate buffer. (b) Tris-HCl buffer. (○) AFB<sub>1</sub>; (+) Tris-diol; (●) aflatoxin Q<sub>1</sub>; (◐) aflatoxin M<sub>1</sub>; (Δ) aflatoxin P<sub>1</sub>.

of the other metabolites were similar using the two buffer systems (fig.2). Paralleling the absence of Tris-diol metabolite from the phosphate incubations there was an increased binding of [<sup>3</sup>H]AFB<sub>1</sub> to the microsomes. That the binding was to microsomal protein was confirmed in some experiments by phenol extraction followed by precipitation of the protein with acetone. Figure 3 shows the inverse relationship between the binding to microsomal protein and the amount of Tris-diol metabolite formed over a range of Tris concentrations.

Authentic AFB<sub>1</sub>-dhd was found to react with certain compounds having an amino group (Tris and iso propylamine) while a related compound not having an amino group (1,3-propane diol) showed no reaction. Some compounds containing amino-groups, e.g., guanine guanosine and guanylic acid showed no reaction, presumably the pK of the amino group is of

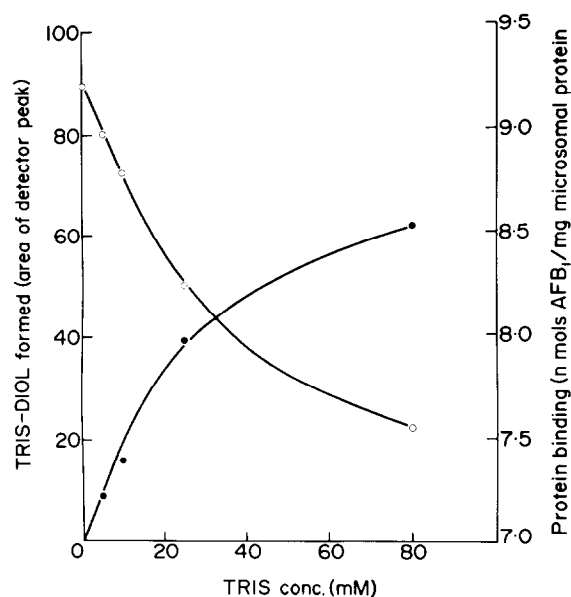


Fig.3. Relationship between Tris-diol formation and protein binding of metabolised AFB<sub>1</sub>. Incubations, preparation of samples, HPLC separations and estimation of protein binding as detailed in section 2. (●) Tris-diol formed; (○) AFB<sub>1</sub> bound.

importance in determining the possibility of reaction at the pH used. The dependance of both the reaction of AFB<sub>1</sub>-dhd with reduced glutathione and the change in ultraviolet absorption spectrum of AFB<sub>1</sub>-dhd with pH is shown in fig.4. Oxidised glutathione yielded similar results. The binding of [<sup>3</sup>H]AFB<sub>1</sub>-dhd to microsomal protein was similarly dependent on pH (fig.5).

The results presented here are probably due to the pH-dependent formation of a dialdehydic phenolate ion from AFB<sub>1</sub>-dhd which then undergoes Schiff's base reaction with amino compounds (fig.4). It has been suggested that aflatoxin B<sub>2a</sub> and AFB<sub>1</sub>-dhd undergo this pH-dependent rearrangement [4,5]. This hypothesis would be supported by the dependence on pH of binding and the bathochromic shift in ultraviolet absorption spectrum reported in the present study. In the absence of acceptor amine, it is possible to reform the AFB<sub>1</sub>-dhd from the dialdehydic phenolate ion by lowering the pH, after reaction with acceptor amine this reversal is not possible (G.E.N., P.J.C., unpublished results). It is probable therefore that the

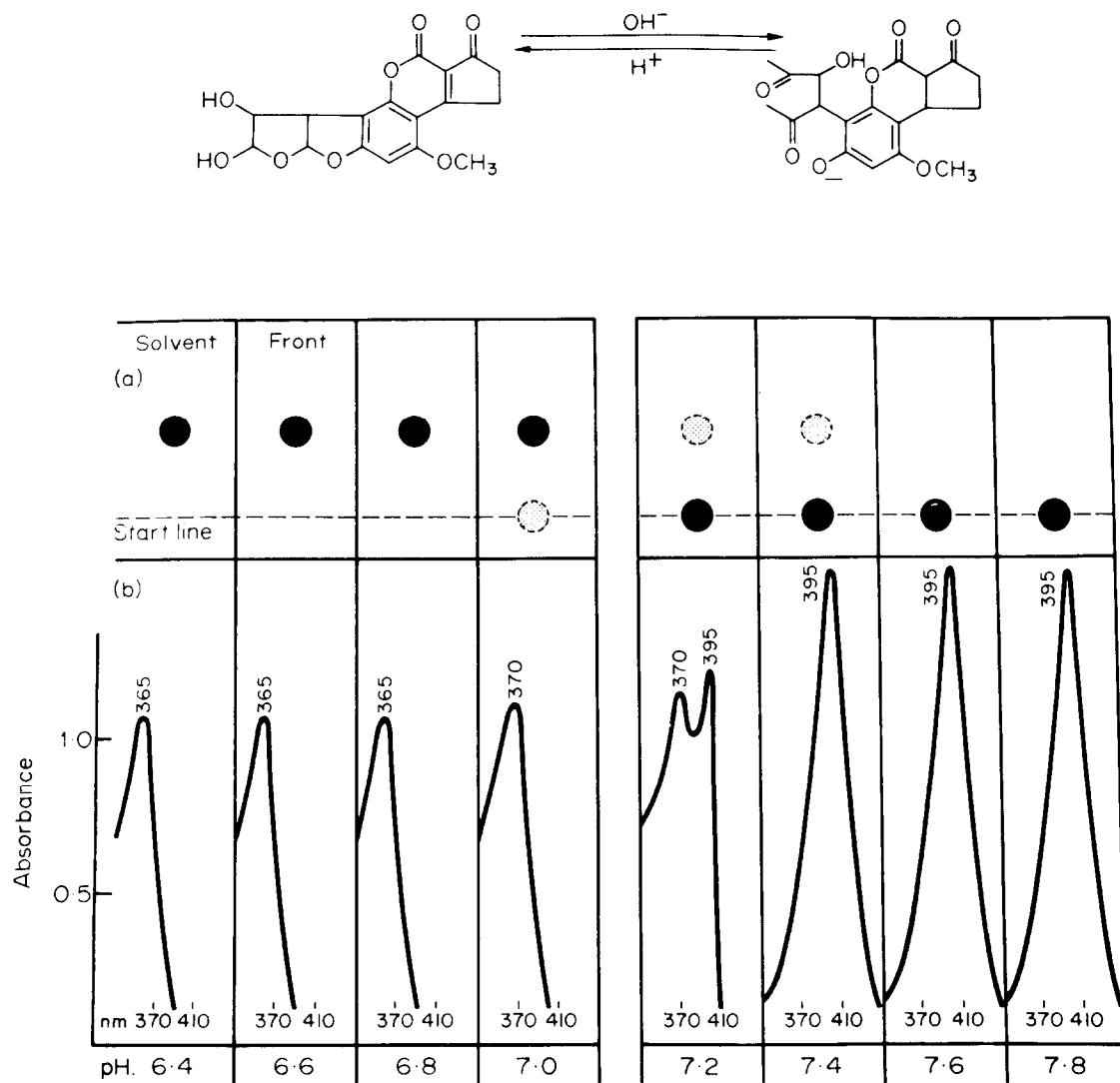


Fig.4. Reaction of AFB<sub>1</sub>-dhd with reduced glutathione. (a) TLC of AFB<sub>1</sub>-dhd after reaction with reduced glutathione at various pH values. Spots represent areas fluorescent under ultraviolet light. Developing solvent chloroform/methanol 1:1 v/v. (b) Ultraviolet absorption spectra of AFB<sub>1</sub>-dhd at several pH values. Suggested pH-dependent molecular rearrangements are also indicated.

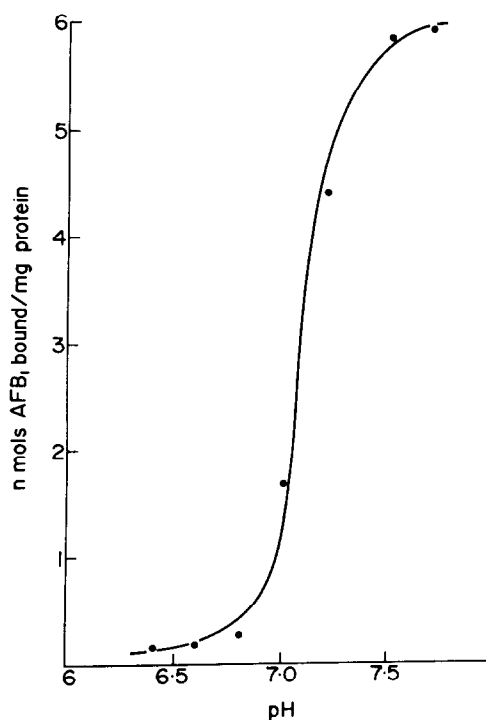


Fig.5. Effect of pH on the binding of AFB<sub>1</sub>-dhd to microsomal protein. [<sup>3</sup>H]AFB<sub>1</sub>-dhd 12.5 nmol (10<sup>4</sup> dpm) and 200  $\mu$ l microsomal suspension in 180 mM KCl (2.1 mg protein) added to 1 ml phosphate buffer at room temperature. After 1 h, diluted 10  $\times$  with distilled water and centrifuged at 100 000  $\times$  g for 1 h. After washing, pellet dissolved in 2 ml 1 M NaOH at 60°C. Aliquots taken for determination of radioactivity.

product of reaction between Tris and dialdehydic phenolate ion form of AFB<sub>1</sub>-dhd is a Schiff's base adduct of these compounds, containing 1 or possibly 2 molecules of Tris per molecule of AFB<sub>1</sub>-dhd. The results of this study would indicate a need for caution in ascribing the formation of adducts of metabolites of xenobiotic compounds with reduced glutathione as being solely due to a reaction of an electrophilic

metabolite of the xenobiotic compound (epoxide?) with the nucleophilic reduced glutathione. Also care should be exercised when using Tris buffers in such systems.

It is interesting that the reaction of AFB<sub>1</sub>-dhd with microsomal protein does not appear to inhibit the mixed function oxidase system. This is shown by the similarities of overall metabolism in phosphate and Tris buffers. Also in other experiments (G.E.N., P.J.C., unpublished) we have found that if microsomal metabolism of AFB<sub>1</sub> is allowed to proceed in duplicate incubations in phosphate buffer one in the presence and the other in the absence of Tris for 15 min, and that Tris is then added to the incubation from which it was previously omitted, the subsequent rate of production of Tris-diol is the same in both incubations.

#### Acknowledgements

The authors would like to thank Mr D. J. Judah for carrying out the protein binding assays and some of the HPLC separations, and Mr J. A. Green for technical assistance.

#### References

- [1] Garner, R. C., Miller, E. C. and Miller, J. A. (1972) *Cancer Res.* 32, 2058–2066.
- [2] Garner, R. C. (1973) *FEBS Lett.* 36, 261–264.
- [3] Lin, J.-K., Kennan, K. A., Miller, E. C. and Miller, J. A. (1978) *Cancer Res.* 38, 2424–2428.
- [4] Patterson, D. S. P. (1976) in: *Biochemical Mechanisms of liver injury* (Slater, T. F. ed) Academic Press, London.
- [5] Swenson, D. G., Miller, J. A. and Miller, E. C. (1975) *Cancer Res.* 35, 3811–3823.
- [6] Neal, G. E. and Colley, P. J. (1978) *Biochem. J.* 174, 839–851.
- [7] Colley, P. J. and Neal, G. E. (1979) *Analyt. Biochem.* in press.